

Diethyl 2-[[3-(2,4,6-trimethylbenzyl)-1-phenylsulfonyl-1*H*-indol-2-yl]methylidene]propanedioate

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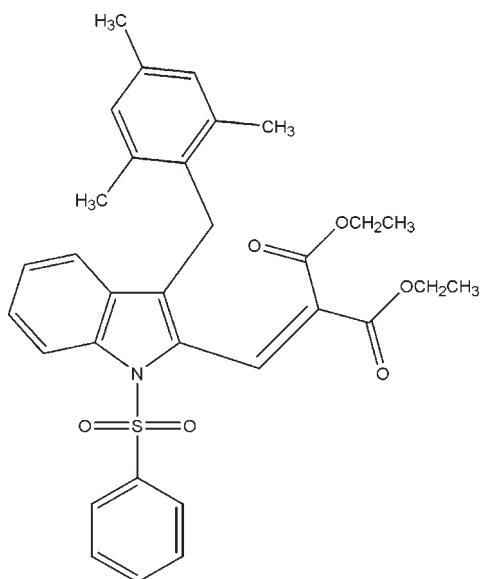
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Key indicators: single-crystal X-ray study; $T = 295$ K; mean $\sigma(\text{C}-\text{C}) = 0.005$ Å; R factor = 0.062; wR factor = 0.211; data-to-parameter ratio = 19.9.

In the title compound, $\text{C}_{32}\text{H}_{33}\text{NO}_6\text{S}$, the indole ring system makes dihedral angles of 62.78 (10) and 80.53 (8)°, respectively, with the phenyl and benzene rings. In the crystal, the molecules are linked through intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds, forming a chain along the a axis. Between the chains, a weak aromatic $\pi-\pi$ stacking interaction [centroid-centroid distance = 3.831 (2) Å] is observed.

Related literature

For the biological activity of indole derivatives, see: Ma *et al.* (2001); Zhao *et al.* (2002); Zhou *et al.* (2006). For related structures, see: Chakkaravarthi *et al.* (2007, 2008).



Experimental

Crystal data

$\text{C}_{32}\text{H}_{33}\text{NO}_6\text{S}$
 $M_r = 559.65$
Triclinic, $P\bar{1}$
 $a = 8.5103$ (4) Å
 $b = 8.9540$ (4) Å
 $c = 19.6546$ (10) Å
 $\alpha = 78.456$ (3)°
 $\beta = 87.236$ (4)°
 $\gamma = 86.736$ (3)°
 $V = 1463.99$ (12) Å³
 $Z = 2$
Mo $K\alpha$ radiation
 $\mu = 0.16$ mm⁻¹
 $T = 295$ K
 $0.22 \times 0.18 \times 0.16$ mm

Data collection

Bruker Kappa APEXII diffractometer
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
 $T_{\min} = 0.967$, $T_{\max} = 0.976$
26660 measured reflections
7349 independent reflections
4328 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.042$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.062$
 $wR(F^2) = 0.211$
 $S = 1.03$
7349 reflections
370 parameters
2 restraints
H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.42$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.31$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C}10-\text{H}10\cdots\text{O}4^i$	0.93	2.43	3.179 (4)	138

Symmetry code: (i) $x + 1, y, z$.

Data collection: *APEX2* (Bruker, 2004); cell refinement: *SAINT* (Bruker, 2004); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: IS2550).

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supplementary materials

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Diethyl 2-{{3-(2,4,6-trimethylbenzyl)-1-phenylsulfonyl-1*H*-indol-2-yl)methylidene}propanedioate

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Comment

Indole derivatives are found to possess anticancer, antimalarial and antihypertensive activities (Ma *et al.*, 2001; Zhou *et al.*, 2006; Zhao *et al.*, 2002). In continuation of our studies in indole derivatives, we report the crystal structure of the title compound, (I). The geometric parameters of (I) (Fig. 1) agree with those in the reported structures (Chakkaravarthi *et al.*, 2007, 2008).

The nine-membered indole ring system forms dihedral angles of 62.78 (10) and 80.53 (8)° with the phenyl ring (C1–C6) and benzene ring (C24–C29), respectively. The torsion angles O2—S1—N1—C7 and O1—S1—N1—C14 [-37.8 (2)° and 62.87 (18)°, respectively] indicate the *syn*-conformation of the sulfonyl moiety. The sum of the bond angles around N1 [342.1 (2)°] indicates that N1 is *sp*³-hybridized.

The molecular structure is stabilized by a weak intramolecular C—H···O interaction and the crystal packing of (I) (Fig. 2) exhibits weak intermolecular C—H···O (Table 1) and π – π interactions [*Cg*···*Cg* (-*x*, -*y*, 1-*z*) distance of 3.831 (2) Å]; *Cg* is the centroid of the C1–C6 ring.

Experimental

To a solution of diethyl-2-((3-(bromomethyl)-1-(phenylsulfonyl)-1*H*-indol-2-yl)methylene)malonate (0.3 g, 0.57 mmol) in dry 1,2-dichloroethane (15 ml), anhydrous ZnBr₂ (0.25 g, 1.11 mmol) and mesitylene (0.19 ml, 1.41 mmol) were added. It was then refluxed for 4 h under N₂ atmosphere. The solvent was removed and the residue was quenched with ice-water (50 ml) containing 1 ml of conc. HCl, extracted with chloroform (2 × 10 ml) and dried (Na₂SO₄). Removal of solvent followed by flash column chromatographic purification (n-hexane/ethyl acetate 98:2) led to the isolation of product as a colourless crystal.

Refinement

H atom attached to C15 was located from a difference Fourier map and refined freely. All other H atoms were positioned geometrically and refined using riding model, with C—H = 0.93 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for aromatic C—H, C—H = 0.97 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for methylene, and C—H = 0.96 Å and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for methyl. C21—C22 and C18—C19 distances were restrained to 1.550 (7) Å.

Figures

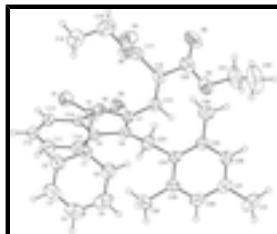


Fig. 1. The molecular structure of (I), with atom labels and 30% probability displacement ellipsoids for non-H atoms.

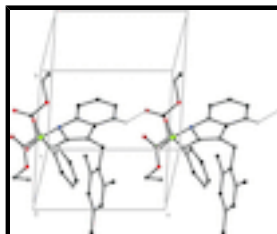


Fig. 2. The packing of (I), viewed down the *c* axis. Hydrogen bonds are shown as dashed lines. H atoms not involved in hydrogen bonding have been omitted.

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$C_{32}H_{33}NO_6S$

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Triclinic, $P\bar{1}$

Hall symbol: -P 1

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$\gamma = 86.736$ (3)°

$V = 1463.99$ (12) Å³

$Z = 2$

$F(000) = 592$

$D_x = 1.270$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 7772 reflections

$\theta = 2.4$ – 24.3 °

$\mu = 0.16$ mm⁻¹

$T = 295$ K

Block, colourless

$0.22 \times 0.18 \times 0.16$ mm

Data collection

Bruker Kappa APEXII
diffractometer

Radiation source: fine-focus sealed tube
graphite

ω and ϕ scans

Absorption correction: multi-scan
(*SADABS*; Sheldrick, 1996)

$T_{\min} = 0.967$, $T_{\max} = 0.976$

26660 measured reflections

7349 independent reflections

4328 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.042$

$\theta_{\max} = 28.5$ °, $\theta_{\min} = 1.1$ °

$h = -11 \rightarrow 10$

$k = -11 \rightarrow 11$

$l = -26 \rightarrow 22$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.062$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.211$	H atoms treated by a mixture of independent and constrained refinement
$S = 1.03$	$w = 1/[\sigma^2(F_o^2) + (0.105P)^2 + 0.4069P]$
7349 reflections	where $P = (F_o^2 + 2F_c^2)/3$
370 parameters	$(\Delta/\sigma)_{\max} < 0.001$
2 restraints	$\Delta\rho_{\max} = 0.42 \text{ e } \text{\AA}^{-3}$
	$\Delta\rho_{\min} = -0.31 \text{ e } \text{\AA}^{-3}$

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	-0.00108 (7)	0.35985 (9)	0.37321 (3)	0.0626 (2)
O1	-0.0678 (2)	0.4651 (3)	0.41334 (11)	0.0837 (6)
O2	-0.0971 (2)	0.2939 (3)	0.33112 (10)	0.0790 (6)
O3	-0.0750 (3)	0.2569 (3)	0.11855 (14)	0.1099 (9)
O4	-0.1997 (3)	0.6190 (3)	0.21742 (15)	0.1050 (8)
O5	-0.1945 (3)	0.4863 (3)	0.08992 (14)	0.1139 (9)
O6	0.0438 (3)	0.6904 (2)	0.18672 (12)	0.0852 (6)
N1	0.1367 (2)	0.4543 (2)	0.31990 (10)	0.0526 (5)
C1	0.1061 (3)	0.2144 (3)	0.42788 (13)	0.0598 (6)
C2	0.1303 (4)	0.2243 (4)	0.49532 (16)	0.0859 (9)
H2	0.0909	0.3089	0.5126	0.103*
C3	0.2130 (5)	0.1083 (5)	0.5370 (2)	0.1025 (12)
H3	0.2274	0.1134	0.5831	0.123*
C4	0.2739 (5)	-0.0135 (5)	0.5118 (2)	0.1069 (13)
H4	0.3293	-0.0919	0.5406	0.128*
C5	0.2542 (6)	-0.0217 (4)	0.4438 (3)	0.1226 (16)
H5	0.2988	-0.1040	0.4261	0.147*
C6	0.1677 (5)	0.0930 (3)	0.40152 (18)	0.0889 (10)
H6	0.1518	0.0873	0.3556	0.107*
C7	0.2050 (3)	0.3920 (3)	0.26267 (12)	0.0489 (5)
C8	0.3634 (3)	0.4001 (2)	0.26078 (12)	0.0487 (5)
C9	0.4031 (3)	0.4704 (2)	0.31715 (12)	0.0490 (5)
C10	0.5466 (3)	0.5094 (3)	0.33863 (15)	0.0623 (6)
H10	0.6403	0.4902	0.3148	0.075*
C11	0.5468 (4)	0.5764 (3)	0.39536 (16)	0.0722 (8)
H11	0.6416	0.6032	0.4102	0.087*
C12	0.4072 (4)	0.6049 (3)	0.43112 (16)	0.0721 (8)
H12	0.4106	0.6489	0.4700	0.087*
C13	0.2652 (3)	0.5699 (3)	0.41068 (14)	0.0653 (7)

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H13	0.1721	0.5906	0.4346	0.078*
C14	0.2644 (3)	0.5025 (3)	0.35296 (12)	0.0503 (5)
C15	0.1071 (3)	0.3562 (3)	0.20960 (13)	0.0563 (6)
C16	-0.0135 (3)	0.4425 (3)	0.18005 (13)	0.0592 (6)
C17	-0.0687 (4)	0.5916 (3)	0.19722 (14)	0.0671 (7)
C18	0.0018 (6)	0.8430 (4)	0.1979 (2)	0.1203 (15)
H19A	-0.1097	0.8655	0.1904	0.144*
H19B	0.0600	0.9163	0.1644	0.144*
C19	0.0364 (8)	0.8585 (6)	0.2684 (3)	0.161 (2)
H66A	-0.0292	0.7935	0.3014	0.241*
H66B	0.0159	0.9626	0.2730	0.241*
H66C	0.1451	0.8294	0.2769	0.241*
C20	-0.1044 (4)	0.4002 (4)	0.12446 (16)	0.0763 (8)
C21	-0.1577 (6)	0.2046 (7)	0.0635 (2)	0.1310 (17)
H21A	-0.1874	0.0999	0.0792	0.157*
H21B	-0.2520	0.2685	0.0512	0.157*
C22	-0.0463 (8)	0.2163 (10)	0.0041 (3)	0.191 (3)
H33A	0.0557	0.1769	0.0198	0.287*
H33B	-0.0402	0.3214	-0.0185	0.287*
H33C	-0.0812	0.1584	-0.0280	0.287*
C23	0.4831 (3)	0.3629 (3)	0.20674 (15)	0.0629 (7)
H23A	0.5870	0.3602	0.2253	0.076*
H23B	0.4785	0.4456	0.1665	0.076*
C24	0.4654 (3)	0.2139 (3)	0.18243 (12)	0.0514 (5)
C25	0.4139 (3)	0.2122 (3)	0.11623 (12)	0.0545 (6)
C26	0.4067 (3)	0.0728 (3)	0.09579 (14)	0.0637 (7)
H26	0.3716	0.0719	0.0518	0.076*
C27	0.4493 (3)	-0.0634 (3)	0.13816 (16)	0.0683 (7)
C28	0.5004 (4)	-0.0589 (3)	0.20301 (16)	0.0722 (8)
H28	0.5302	-0.1500	0.2324	0.087*
C29	0.5091 (3)	0.0768 (3)	0.22612 (13)	0.0606 (6)
C30	0.3701 (4)	0.3567 (3)	0.06561 (15)	0.0847 (9)
H30A	0.4612	0.4166	0.0538	0.127*
H30B	0.2894	0.4142	0.0865	0.127*
H30C	0.3319	0.3315	0.0244	0.127*
C31	0.4440 (5)	-0.2144 (4)	0.1150 (2)	0.1061 (13)
H31A	0.5475	-0.2632	0.1164	0.159*
H31B	0.4078	-0.1970	0.0685	0.159*
H31C	0.3732	-0.2789	0.1456	0.159*
C32	0.5710 (5)	0.0727 (5)	0.29715 (17)	0.0964 (11)
H32A	0.5875	-0.0314	0.3207	0.145*
H32B	0.4960	0.1244	0.3236	0.145*
H32C	0.6690	0.1227	0.2924	0.145*
H15	0.143 (4)	0.268 (4)	0.1932 (16)	0.085 (9)*

Atomic displacement parameters (\AA^2)

U^{11}

U^{22}

U^{33}

U^{12}

U^{13}

U^{23}

S1	0.0375 (3)	0.0952 (5)	0.0569 (4)	0.0061 (3)	-0.0024 (3)	-0.0216 (3)
O1	0.0545 (11)	0.1244 (17)	0.0742 (13)	0.0296 (11)	0.0046 (9)	-0.0356 (12)
O2	0.0433 (10)	0.1292 (17)	0.0664 (11)	-0.0156 (10)	-0.0065 (8)	-0.0196 (12)
O3	0.116 (2)	0.123 (2)	0.1081 (18)	0.0088 (16)	-0.0507 (16)	-0.0572 (16)
O4	0.0776 (16)	0.0955 (16)	0.132 (2)	0.0165 (13)	0.0206 (15)	-0.0106 (15)
O5	0.114 (2)	0.134 (2)	0.0925 (17)	0.0064 (17)	-0.0521 (16)	-0.0129 (16)
O6	0.0803 (14)	0.0767 (13)	0.1012 (16)	-0.0048 (11)	-0.0096 (12)	-0.0222 (12)
N1	0.0419 (10)	0.0665 (12)	0.0531 (11)	0.0079 (9)	-0.0069 (8)	-0.0224 (9)
C1	0.0464 (13)	0.0751 (16)	0.0592 (14)	-0.0097 (12)	-0.0018 (11)	-0.0144 (12)
C2	0.082 (2)	0.112 (2)	0.0647 (18)	0.0148 (19)	-0.0113 (16)	-0.0244 (18)
C3	0.100 (3)	0.126 (3)	0.076 (2)	0.005 (2)	-0.025 (2)	-0.003 (2)
C4	0.112 (3)	0.079 (2)	0.117 (3)	-0.005 (2)	-0.026 (3)	0.016 (2)
C5	0.171 (5)	0.0572 (19)	0.136 (4)	0.013 (2)	-0.016 (3)	-0.014 (2)
C6	0.126 (3)	0.0638 (17)	0.080 (2)	-0.0078 (18)	-0.013 (2)	-0.0185 (16)
C7	0.0446 (12)	0.0520 (12)	0.0521 (12)	0.0033 (9)	-0.0016 (10)	-0.0168 (10)
C8	0.0460 (12)	0.0451 (11)	0.0557 (13)	0.0033 (9)	-0.0004 (10)	-0.0132 (10)
C9	0.0453 (12)	0.0443 (11)	0.0578 (13)	0.0030 (9)	-0.0075 (10)	-0.0111 (10)
C10	0.0492 (14)	0.0630 (15)	0.0762 (17)	-0.0023 (11)	-0.0088 (12)	-0.0160 (13)
C11	0.0664 (18)	0.0683 (16)	0.087 (2)	-0.0012 (13)	-0.0280 (15)	-0.0228 (15)
C12	0.081 (2)	0.0662 (16)	0.0774 (18)	0.0166 (14)	-0.0329 (16)	-0.0323 (14)
C13	0.0630 (16)	0.0722 (16)	0.0663 (16)	0.0201 (13)	-0.0149 (13)	-0.0305 (13)
C14	0.0471 (12)	0.0492 (12)	0.0571 (13)	0.0112 (10)	-0.0116 (10)	-0.0175 (10)
C15	0.0494 (13)	0.0685 (15)	0.0556 (14)	-0.0023 (11)	0.0002 (11)	-0.0243 (12)
C16	0.0516 (14)	0.0766 (16)	0.0496 (13)	-0.0056 (12)	-0.0022 (11)	-0.0121 (12)
C17	0.0627 (17)	0.0747 (17)	0.0579 (15)	0.0066 (14)	-0.0037 (13)	-0.0016 (13)
C18	0.129 (4)	0.080 (2)	0.152 (4)	0.010 (2)	-0.020 (3)	-0.025 (3)
C19	0.221 (7)	0.123 (4)	0.158 (5)	0.045 (4)	-0.067 (5)	-0.075 (3)
C20	0.0659 (18)	0.100 (2)	0.0638 (17)	-0.0061 (16)	-0.0127 (14)	-0.0151 (16)
C21	0.119 (4)	0.178 (5)	0.116 (3)	-0.015 (3)	-0.037 (3)	-0.069 (3)
C22	0.170 (6)	0.307 (9)	0.127 (4)	-0.039 (6)	-0.001 (4)	-0.109 (6)
C23	0.0544 (14)	0.0630 (14)	0.0751 (17)	-0.0084 (12)	0.0160 (12)	-0.0250 (13)
C24	0.0454 (12)	0.0514 (12)	0.0561 (13)	0.0037 (10)	0.0065 (10)	-0.0116 (10)
C25	0.0534 (13)	0.0565 (13)	0.0496 (13)	0.0071 (10)	0.0074 (10)	-0.0058 (10)
C26	0.0599 (15)	0.0753 (17)	0.0586 (15)	0.0010 (13)	0.0005 (12)	-0.0217 (13)
C27	0.0668 (17)	0.0553 (14)	0.0840 (19)	0.0031 (12)	0.0043 (14)	-0.0204 (14)
C28	0.0753 (19)	0.0520 (14)	0.082 (2)	0.0174 (13)	-0.0007 (15)	-0.0013 (13)
C29	0.0551 (14)	0.0650 (15)	0.0589 (15)	0.0131 (12)	-0.0055 (11)	-0.0092 (12)
C30	0.106 (3)	0.0739 (18)	0.0615 (17)	0.0176 (17)	0.0107 (16)	0.0080 (14)
C31	0.113 (3)	0.0680 (19)	0.146 (3)	-0.0031 (19)	0.008 (3)	-0.046 (2)
C32	0.096 (3)	0.117 (3)	0.074 (2)	0.029 (2)	-0.0270 (18)	-0.0164 (19)

Geometric parameters (Å, °)

S1—O2	1.420 (2)	C15—H15	0.94 (3)
S1—O1	1.424 (2)	C16—C17	1.487 (4)
S1—N1	1.682 (2)	C16—C20	1.488 (4)
S1—C1	1.755 (3)	C18—C19	1.465 (5)
O3—C20	1.320 (4)	C18—H19A	0.9700
O3—C21	1.483 (4)	C18—H19B	0.9700

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O4—C17	1.194 (4)	C19—H66A	0.9600
O5—C20	1.190 (4)	C19—H66B	0.9600
O6—C17	1.321 (4)	C19—H66C	0.9600
O6—C18	1.449 (4)	C21—C22	1.458 (5)
N1—C14	1.422 (3)	C21—H21A	0.9700
N1—C7	1.438 (3)	C21—H21B	0.9700
C1—C6	1.363 (4)	C22—H33A	0.9600
C1—C2	1.372 (4)	C22—H33B	0.9600
C2—C3	1.370 (5)	C22—H33C	0.9600
C2—H2	0.9300	C23—C24	1.522 (3)
C3—C4	1.354 (6)	C23—H23A	0.9700
C3—H3	0.9300	C23—H23B	0.9700
C4—C5	1.371 (6)	C24—C29	1.392 (3)
C4—H4	0.9300	C24—C25	1.396 (3)
C5—C6	1.386 (5)	C25—C26	1.391 (3)
C5—H5	0.9300	C25—C30	1.506 (4)
C6—H6	0.9300	C26—C27	1.373 (4)
C7—C8	1.353 (3)	C26—H26	0.9300
C7—C15	1.460 (3)	C27—C28	1.375 (4)
C8—C9	1.441 (3)	C27—C31	1.514 (4)
C8—C23	1.508 (3)	C28—C29	1.387 (4)
C9—C14	1.391 (3)	C28—H28	0.9300
C9—C10	1.396 (3)	C29—C32	1.509 (4)
C10—C11	1.368 (4)	C30—H30A	0.9600
C10—H10	0.9300	C30—H30B	0.9600
C11—C12	1.389 (4)	C30—H30C	0.9600
C11—H11	0.9300	C31—H31A	0.9600
C12—C13	1.364 (4)	C31—H31B	0.9600
C12—H12	0.9300	C31—H31C	0.9600
C13—C14	1.388 (3)	C32—H32A	0.9600
C13—H13	0.9300	C32—H32B	0.9600
C15—C16	1.332 (4)	C32—H32C	0.9600
O2—S1—O1	120.54 (13)	C19—C18—H19B	109.4
O2—S1—N1	106.83 (11)	H19A—C18—H19B	108.0
O1—S1—N1	105.76 (13)	C18—C19—H66A	109.5
O2—S1—C1	109.32 (13)	C18—C19—H66B	109.5
O1—S1—C1	108.79 (13)	H66A—C19—H66B	109.5
N1—S1—C1	104.36 (11)	C18—C19—H66C	109.5
C20—O3—C21	116.6 (3)	H66A—C19—H66C	109.5
C17—O6—C18	117.1 (3)	H66B—C19—H66C	109.5
C14—N1—C7	106.24 (18)	O5—C20—O3	124.2 (3)
C14—N1—S1	115.80 (16)	O5—C20—C16	123.4 (3)
C7—N1—S1	120.08 (16)	O3—C20—C16	112.5 (3)
C6—C1—C2	120.7 (3)	C22—C21—O3	105.9 (4)
C6—C1—S1	118.6 (2)	C22—C21—H21A	110.5
C2—C1—S1	120.7 (2)	O3—C21—H21A	110.5
C3—C2—C1	119.4 (3)	C22—C21—H21B	110.5
C3—C2—H2	120.3	O3—C21—H21B	110.5
C1—C2—H2	120.3	H21A—C21—H21B	108.7

C4—C3—C2	120.6 (4)	C21—C22—H33A	109.5
C4—C3—H3	119.7	C21—C22—H33B	109.5
C2—C3—H3	119.7	H33A—C22—H33B	109.5
C3—C4—C5	120.2 (4)	C21—C22—H33C	109.5
C3—C4—H4	119.9	H33A—C22—H33C	109.5
C5—C4—H4	119.9	H33B—C22—H33C	109.5
C4—C5—C6	119.7 (4)	C8—C23—C24	116.5 (2)
C4—C5—H5	120.1	C8—C23—H23A	108.2
C6—C5—H5	120.1	C24—C23—H23A	108.2
C1—C6—C5	119.3 (3)	C8—C23—H23B	108.2
C1—C6—H6	120.4	C24—C23—H23B	108.2
C5—C6—H6	120.4	H23A—C23—H23B	107.3
C8—C7—N1	109.75 (19)	C29—C24—C25	119.5 (2)
C8—C7—C15	128.0 (2)	C29—C24—C23	119.0 (2)
N1—C7—C15	121.3 (2)	C25—C24—C23	121.5 (2)
C7—C8—C9	107.7 (2)	C26—C25—C24	118.9 (2)
C7—C8—C23	128.9 (2)	C26—C25—C30	119.0 (2)
C9—C8—C23	123.0 (2)	C24—C25—C30	122.1 (2)
C14—C9—C10	119.5 (2)	C27—C26—C25	122.4 (3)
C14—C9—C8	108.2 (2)	C27—C26—H26	118.8
C10—C9—C8	132.3 (2)	C25—C26—H26	118.8
C11—C10—C9	118.7 (3)	C26—C27—C28	117.6 (2)
C11—C10—H10	120.6	C26—C27—C31	122.1 (3)
C9—C10—H10	120.6	C28—C27—C31	120.2 (3)
C10—C11—C12	120.8 (3)	C27—C28—C29	122.3 (2)
C10—C11—H11	119.6	C27—C28—H28	118.9
C12—C11—H11	119.6	C29—C28—H28	118.9
C13—C12—C11	121.7 (3)	C28—C29—C24	119.3 (2)
C13—C12—H12	119.1	C28—C29—C32	119.1 (3)
C11—C12—H12	119.1	C24—C29—C32	121.6 (3)
C12—C13—C14	117.7 (3)	C25—C30—H30A	109.5
C12—C13—H13	121.1	C25—C30—H30B	109.5
C14—C13—H13	121.1	H30A—C30—H30B	109.5
C13—C14—C9	121.5 (2)	C25—C30—H30C	109.5
C13—C14—N1	130.4 (2)	H30A—C30—H30C	109.5
C9—C14—N1	108.11 (19)	H30B—C30—H30C	109.5
C16—C15—C7	126.1 (2)	C27—C31—H31A	109.5
C16—C15—H15	120.2 (19)	C27—C31—H31B	109.5
C7—C15—H15	113.4 (19)	H31A—C31—H31B	109.5
C15—C16—C17	124.2 (2)	C27—C31—H31C	109.5
C15—C16—C20	122.8 (3)	H31A—C31—H31C	109.5
C17—C16—C20	113.0 (2)	H31B—C31—H31C	109.5
O4—C17—O6	124.2 (3)	C29—C32—H32A	109.5
O4—C17—C16	123.9 (3)	C29—C32—H32B	109.5
O6—C17—C16	112.0 (2)	H32A—C32—H32B	109.5
O6—C18—C19	111.3 (4)	C29—C32—H32C	109.5
O6—C18—H19A	109.4	H32A—C32—H32C	109.5
C19—C18—H19A	109.4	H32B—C32—H32C	109.5
O6—C18—H19B	109.4		

supplementary materials

O2—S1—N1—C14	-167.56 (16)	C7—N1—C14—C13	-179.8 (3)
O1—S1—N1—C14	62.87 (18)	S1—N1—C14—C13	-43.7 (3)
C1—S1—N1—C14	-51.82 (18)	C7—N1—C14—C9	0.0 (2)
O2—S1—N1—C7	-37.8 (2)	S1—N1—C14—C9	136.08 (18)
O1—S1—N1—C7	-167.42 (17)	C8—C7—C15—C16	-124.3 (3)
C1—S1—N1—C7	77.89 (19)	N1—C7—C15—C16	42.8 (4)
O2—S1—C1—C6	40.3 (3)	C7—C15—C16—C17	-0.3 (4)
O1—S1—C1—C6	173.8 (2)	C7—C15—C16—C20	177.9 (2)
N1—S1—C1—C6	-73.7 (3)	C18—O6—C17—O4	-2.7 (5)
O2—S1—C1—C2	-141.1 (2)	C18—O6—C17—C16	176.6 (3)
O1—S1—C1—C2	-7.6 (3)	C15—C16—C17—O4	-122.4 (3)
N1—S1—C1—C2	104.9 (3)	C20—C16—C17—O4	59.2 (4)
C6—C1—C2—C3	-2.1 (5)	C15—C16—C17—O6	58.3 (4)
S1—C1—C2—C3	179.4 (3)	C20—C16—C17—O6	-120.1 (3)
C1—C2—C3—C4	1.6 (6)	C17—O6—C18—C19	94.3 (5)
C2—C3—C4—C5	0.4 (7)	C21—O3—C20—O5	1.9 (6)
C3—C4—C5—C6	-1.9 (7)	C21—O3—C20—C16	-178.8 (3)
C2—C1—C6—C5	0.6 (5)	C15—C16—C20—O5	-167.4 (3)
S1—C1—C6—C5	179.2 (3)	C17—C16—C20—O5	11.1 (4)
C4—C5—C6—C1	1.4 (7)	C15—C16—C20—O3	13.4 (4)
C14—N1—C7—C8	0.4 (3)	C17—C16—C20—O3	-168.2 (3)
S1—N1—C7—C8	-133.46 (18)	C20—O3—C21—C22	97.3 (6)
C14—N1—C7—C15	-168.9 (2)	C7—C8—C23—C24	-46.2 (4)
S1—N1—C7—C15	57.3 (3)	C9—C8—C23—C24	141.6 (2)
N1—C7—C8—C9	-0.6 (3)	C8—C23—C24—C29	-74.8 (3)
C15—C7—C8—C9	167.8 (2)	C8—C23—C24—C25	108.4 (3)
N1—C7—C8—C23	-173.7 (2)	C29—C24—C25—C26	0.4 (4)
C15—C7—C8—C23	-5.4 (4)	C23—C24—C25—C26	177.2 (2)
C7—C8—C9—C14	0.5 (3)	C29—C24—C25—C30	-177.9 (2)
C23—C8—C9—C14	174.2 (2)	C23—C24—C25—C30	-1.1 (4)
C7—C8—C9—C10	-178.5 (2)	C24—C25—C26—C27	-0.5 (4)
C23—C8—C9—C10	-4.9 (4)	C30—C25—C26—C27	177.9 (3)
C14—C9—C10—C11	1.0 (4)	C25—C26—C27—C28	0.1 (4)
C8—C9—C10—C11	-180.0 (2)	C25—C26—C27—C31	-178.8 (3)
C9—C10—C11—C12	0.2 (4)	C26—C27—C28—C29	0.3 (4)
C10—C11—C12—C13	-1.1 (5)	C31—C27—C28—C29	179.3 (3)
C11—C12—C13—C14	0.8 (4)	C27—C28—C29—C24	-0.3 (4)
C12—C13—C14—C9	0.4 (4)	C27—C28—C29—C32	-178.2 (3)
C12—C13—C14—N1	-179.9 (2)	C25—C24—C29—C28	0.0 (4)
C10—C9—C14—C13	-1.3 (4)	C23—C24—C29—C28	-176.9 (2)
C8—C9—C14—C13	179.5 (2)	C25—C24—C29—C32	177.8 (3)
C10—C9—C14—N1	178.9 (2)	C23—C24—C29—C32	0.9 (4)
C8—C9—C14—N1	-0.3 (3)		

Hydrogen-bond geometry (\AA , $^\circ$)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C10—H10 \cdots O4 ⁱ	0.93	2.43	3.179 (4)	138
C13—H13 \cdots O1	0.93	2.48	3.030 (4)	118

Symmetry codes: (i) $x+1, y, z$.

Fig. 1

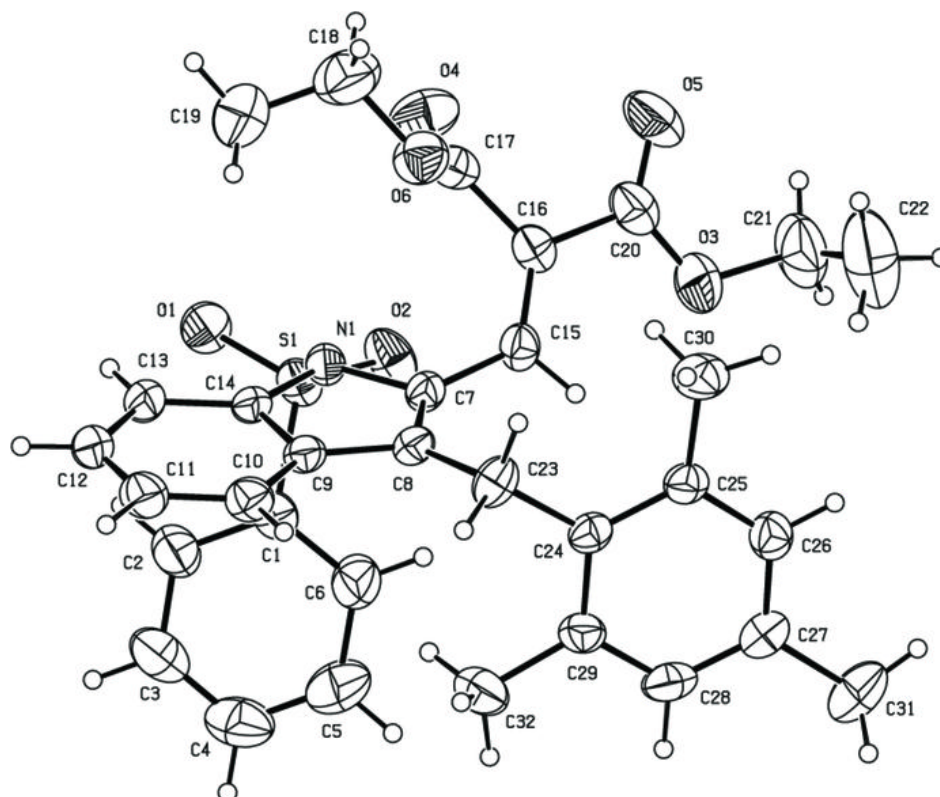


Fig. 2

